

- Supporting information -

Influence of Bi doping on physical properties of lead halide
perovskites: a comparative first-principles study between CsPbI₃
and CsPbBr₃

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Table S1

The cell-size dependence of the formation energy of $\text{Bi}_{\text{pb}}^{1+}$ in $\alpha\text{-CsPbI}_3$ using the PBE without the SOC effect. The value is presented as the relative value (ΔE^f) relative to the one using a $5\times 5\times 5$ supercell.

Supercell size	$\Delta E^f(\text{Bi}_{\text{pb}}^{1+})$ (eV)
$3\times 3\times 3$	-0.0745
$4\times 4\times 4$	0.0006
$5\times 5\times 5$	0.0

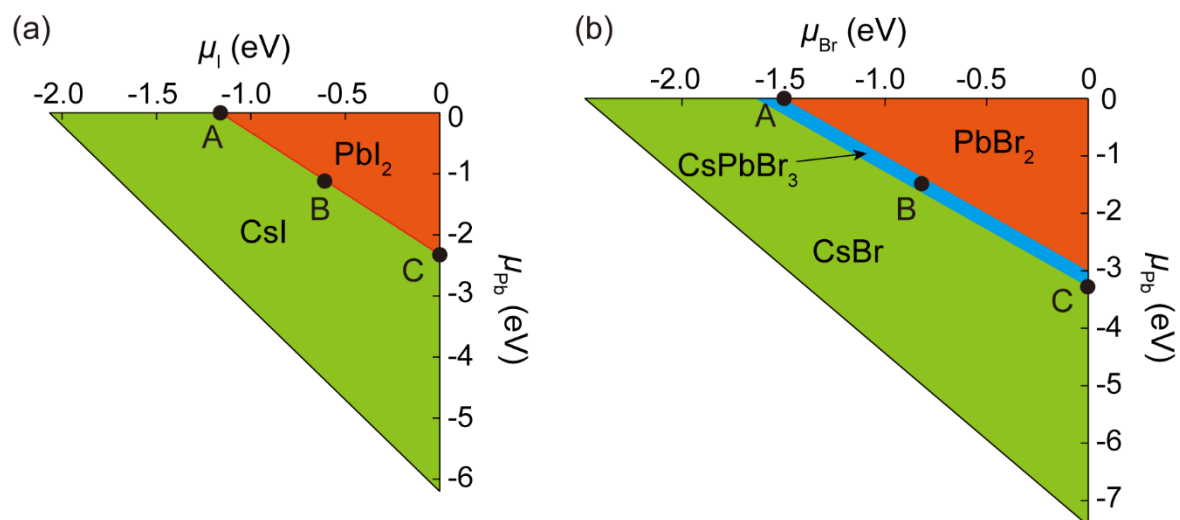


Fig S1. Stability regions of different compounds with respect to X and Pb chemical potentials in CsPbX_3 [$X = \text{I}$ for (a) and Br for (b)]. Chemical potentials of each element referenced to the energy of their own stable phases are listed in Table S2 for three representative points A (X-poor), B (intermediate), and C (X-rich).

Table S2

The chemical potentials of each element in eV depending on the growth conditions shown in Fig. S1. The values for Bi-rich are used for calculating the formation energy of BiPb.

		α -CsPbI ₃	δ -CsPbBr ₃
A	μ_{Cs}	-2.72	-2.93
	μ_{Pb}	0.00	0.00
	μ_{X}	-1.16	-1.50
B	μ_{Cs}	-3.30	-3.54
	μ_{Pb}	-1.16	-1.64
	μ_{X}	-0.58	-0.75
C	μ_{Cs}	-3.88	-4.16
	μ_{Pb}	-2.32	-3.28
	μ_{X}	0.00	0.00
Bi-rich	μ_{Cs}	-3.21	-3.50
	μ_{Pb}	-1.00	-1.56
	μ_{X}	-0.66	-0.79

Calculation of the equilibrium Fermi level

For the calculation of the equilibrium Fermi level (or the equilibrium concentration of free carriers, the native acceptors, and ionized Bi_{Pb}) at a given temperature, we applied the charge neutrality condition which is given by $n = [\text{Bi}_{\text{Pb}}^{1+}]$ if the compensation by the native acceptors is negligible.

Using Eqs. 4 and 5 in the text, this neutrality condition can be expressed by $N_{\text{C}} \exp\left(\frac{E_{\text{F}} - \text{CBM}}{kT}\right) = \frac{[\text{Bi}_{\text{Pb}}^{\text{tot}}]}{1 + 2 \exp\left\{\frac{E_{\text{F}} - \varepsilon(1+/0)}{kT}\right\}}$. The consideration of effects of the native acceptors modifies the neutrality

condition as $n + [V_{\text{Cs}}^{1-}] + 2 \times [V_{\text{Pb}}^{2-}] + [I_1^{1-}] = [\text{Bi}_{\text{Pb}}^{1+}]$, which can be expressed as

$$N_{\text{C}} \exp\left(\frac{E_{\text{F}} - \text{CBM}}{kT}\right) + N_{\text{Cs}} \exp\left\{-\frac{E^f(V_{\text{Cs}}^{1-})}{kT}\right\} + 2 \times N_{\text{Pb}} \exp\left\{-\frac{E^f(V_{\text{Pb}}^{2-})}{kT}\right\} + N_{\text{Cs}} \exp\left\{-\frac{E^f(V_{\text{Cs}}^{1-})}{kT}\right\} =$$

$\frac{[\text{Bi}_{\text{Pb}}^{\text{tot}}]}{1 + 2 \exp\left\{\frac{E_{\text{F}} - \varepsilon(1+/0)}{kT}\right\}}$ by using Eqs. 4-6. For a given $[\text{Bi}_{\text{Pb}}^{\text{tot}}]$, we determined the equilibrium Fermi level

by evaluating E_{F} to solve the above neutrality equations. For instance, neglecting the compensation by native acceptors, the equilibrium Fermi level for CsPbBr_3 including $[\text{Bi}_{\text{Pb}}^{\text{tot}}] = 10^{20} \text{ cm}^{-3}$ is $\text{CBM} - 0.25 \text{ eV}$, which leads to the equal concentration ($\sim 10^{14} \text{ cm}^{-3}$) of free electrons and $[\text{Bi}_{\text{Pb}}^{1+}]$.