Supporting Information

Double-layer Graphene Outperforming Monolayer as Catalyst on Silicon Photocathode for Hydrogen Production

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Figure S1. Schematic illustration and photograph images of hydrogen evolution reaction on graphene/Si Photocathode.



Figure S2. (a-d) Spatially resolved Raman mapping plotted with I(2D) of mono-, bi-, triand multilayer graphene, respectively. Scan size: 20 μ m x 20 μ m. Each Raman map shows uniform surface morphology.



Figure S3. Photoelectrochemical performance of catalysts on p-Si. ABPE indicates applied bias photon-to-current efficiency. 'Bare' indicates the performance of the bare electrode without deposit of the catalyst. Detail information about references can also be shown in Table S1.

Table S1. Summary of experimental data for non-noble metal/metal oxide or carbon-based

Electrode	Onset potential [E (V _{RHE} at -1 mA/cm ²]	E (V_{RHE}) at -5 mA/cm ²	E (V _{RHE}) at -10 mA/cm ²	ABPE (%)	References	
Bare Si	-0.13	-0.27	-0.35	0.005		
Pristine graphene 1- Si	-0.03	-0.09	-0.16	0.04		
Gr 2-Si	0.05	-0.04	-0.11	0.05		
Gr 3-Si	0.04	-0.06	-0.13	0.04	 .	
Gr 4-Si	0.02	-0.10	-0.18	0.03	- This study	
Gr 5-Si	-0.10	-0.22	-0.29	0.005		
Plasma Gr 1-Si	0.11	0.01	-0.04	0.14		
Plasma Gr 2-Si	0.15	0.07	0.01	0.32		
Plasma Gr 3-Si	0.15	0.03	-0.06	0.24		
Plasma Gr 4-Si	0.12	0.00	-0.08	0.15		
Plasma Gr 5-Si	-0.03	-0.16	-0.25	0.01		
Bare p-Si	-0.41	-0.55	-0.60	N/A	Nano Lett. 2011, 12, 298	
Bare p-Si	-0.35	-0.52	-0.60	N/A	<i>Energy Environ. Sci.,</i> 2011, 4 , 1690-1694	
Bare p-Si	-0.03	-0.17	N/A (saturated at -9 mA/cm ²)	N/A	ACS Appl. Mater. Interfaces. 2013, 5 , 1961	
rGO on p-Si	+0.08	-0.05	-0.14	0.005		
Bare	-0.51	-0.64	-0.77	N/A	Nat Mator 2011 10 424	
Mo ₃ S ₄ on p-Si	+0.12	+0.07	-0.01	~0.2	_ Nat. Mater. 2011, 10, 434	
CuW ₃ S ₄ on p-Si	-0.16	-0.22	-0.29	N/A (~≤0.001)		
CuMo ₃ S ₄ on p-Si	+0.12	+0.06	0.00	~0.2	_	
W ₃ S ₄ on p-Si	-0.18	-0.25	-0.31	N/A (~≤0.001)	<i>J. Photon. Energy</i> 2012, 2 , 026001	
CoW ₃ S ₄ on p-Si (1st scan)	-0.12	-0.22	-0.3	N/A (~≤0.001)	-	
CoMo ₃ S ₄ on p-Si (1st scan)	-0.02	-0.08	-0.15	N/A (~≤0.001)	-	
Bare p-Si	-0.50	N/A	N/A	N/A	Chem. Eur. J. 2012. 18.	

catalysts on p-Si electrodes.

MoS ₂ on Si	-0.06	-0.25	-0.35	0.04	13994	
Bare p-Si	-0.25	-0.35	-0.45	N/A	J. Mater. Chem. A 2014, 2,	
g-C ₃ N ₄	0	-0.15	-0.25	~0.2	- 12697	
Bare p-Si	-0.34	-0.47	-0.53	N/A	Thin Solid Films 2016,	
1.4 nm Al ₂ O ₃	-0.21	-0.33	-0.4	N/A (~≤0.001)	616 , 550-554	
Bare p-Si	-0.05	-0.28	-0.38	N/A	Nano Res. 2015, 8 (5),	
5 nm Ni/15 nm Ti/p- Si	+0.22	+0.12	0.08	~1.44	1577	
Bare p-Si	-0.7	-0.9	-1.0	N/A		
TiO ₂ film/p-Si	-0.4	-0.5	-0.6	N/A	J. Mater. Chem. A 2016, 4 , 9477	
TiO ₂ nanorod/p-Si	-0.05	-0.13	-0.25	N/A (~≤0.002)	_	
SrTiO ₃ on p-Si	+0.42	+0.25	+0.01	~1.56	<i>Nat. Nanotech.</i> 2015, 10 , 84	
WS ₂ on p-Si	+0.12	+0.03	-0.04	~0.28	ACS Appl. Mater. Interfaces. 2014, 6 , 10408	
WS ₃ on p-Si	+0.36	+0.27	+0.20	~2		

Values were measured and extrapolated by our group referring to the figures and data from other papers.



Figure S4. Photoelectrochemical response of graphene on silicon. Saturation current density (Saturation current density) from photocurrent density-potential (J-E) curves of different number of graphene layers on lightly boron-doped planar p-Si electrode.



Figure S5. Stability testing of bare Si and graphene layers on Si photocathodes, showing the change in normalized photocurrent density at 0 V vs. RHE of each photoelectrodes as a function of operation time.



Figure S6. Dark current densities of graphene layers on normal B-doped Si electrodes.



Figure S7. Electrochemical activity of graphene layers on glassy carbon electrode. Cyclivoltammetry (CV) curve of bare glassy carbon, compared to graphene layers on glassy carbon from a rotating disk electrode system. CV data were corrected with iR compensation.



Figure S8. Equivalent circuit to analyze the electrochemical impedance spectroscopy data. $R_{overall}$ is the overall series resistance of the circuit, R_1 is the charge transfer resistance of the double layer at interface between electrode and electrolyte, C_1 is the capacitance phase element of the double layer at interface between electrode and electrolyte, R_2 is the charge transfer resistance of the depletion layer in graphene/Si and C_2 is the capacitance phase element of the depletion layer in graphene/Si.

Sample	R _{overall}	R_2	C_2	R_1	C_1
	(Ω .cm²)	(Ω.cm²)	(F/cm²)	(Ω.cm²)	(F/cm²)
Bare Si	11.65	3083	2.081E-06	-	-
Pristine Gr 1L	12.76	10.51	1.49E-06	74.39	3.439E-06
Pristine Gr 2L	11.61	8.672	2.043E-06	33.34	7.68E-06
Pristine Gr 3L	22.33	16.86	5.081E-07	49.49	<i>3.14E-06</i>
Pristine Gr 4L	20.32	19.906	1.54E-06	88.26	3.918E-06
Pristine Gr 5L	24.36	17.461	2.93E-06	218.3	4.377E-06
Plasma Gr 1L	11.63	9.700	2.38E-06	43.03	1.002E-05
Plasma Gr 2L	11.74	8.588	2.057E-06	26.11	1.16E-05
Plasma Gr 3L	17.66	12.64	1.17E-06	40.15	6.638E-06
Plasma Gr 4L	17.58	11.38	1.13E-06	46.58	6.081E-06
Plasma Gr 5L	23.71	19.372	1.360E-06	86.74	2.44E-06

Table S2. Results obtained from the fitting of the EIS data.



Figure S9. Surface characterization of the different number of plasma-treated graphene layers. a-d, Atomic force microscopy (AFM) images of different number of plasma-treated graphene layers. **e,** Raman spectra of plasma-treated. **f,** Transmittance of plasma-treated graphene specimens on polyethylene terephthalate substrate.



Figure S10. Comparison of electrochemical data between pristine graphene and plasmatreated graphene. a, Potential vs. RHE at -5 mA/cm² of different number of graphene layers. b, Flatband potential of pristine graphene and plasma-treated graphene. c, Work function difference between graphene and the Si.



Figure S11. Comparison of electrochemical data between pristine graphene and plasmatreated graphene. Charge transfer resistance of different number of graphene layers. \mathbf{a} , R_1 is the charge transfer resistance of the electrical double layer at interface between electrode and electrolyte. \mathbf{b} , R_2 is the charge transfer resistance of the depletion layer in graphene/Si.



Figure S12. Computational modelling of p-Si/graphene junction. a, Unit cell structure for the modelling within the periodic boundary condition. Red boxes in the top views are unit cells of the reconstructed Si(100) surface and graphene. **b,** xy-plane averaged electron concentration change after junction formation. **c-i (c-ii),** Band structures of graphene monolayer after (before) formation of junction, and the color intensity on c-i shows the weight of C atoms.